

CORIAN[®] SOLID SURFACE CHEMICAL RESISTANCE

INTRODUCTION

This technical bulletin discusses the chemical resistance of Corian[®] Solid Surface. Corian[®] Solid Surface with Resilience Technology[™] offers similar chemical resistance, but renewing the surface may be easier with some chemicals. Chemical resistance is evaluated by placing a material on Corian[®] Solid Surface and covering it for 16 hours. Time of exposure is an important factor; prompt removal of chemicals will prevent most damage.

The concentration tested is listed where applicable, unless specified the chemical is a solution in water. Use caution if using higher concentrations as they may increase the likelihood of damage. Concentrations reported

acetic acid (10%)
acetone
ammonium hydroxide (<28%) (ammonia in water)
amyl acetate
amyl alcohol
aromatic ammonia (smelling salts)
ball point pen ink
benzene†
bleach (household type)
blood
butanol (butyl alcohol)
calcium thiocyanate (78%)
carbon disulfide
carbon tetrachloride
cigarette (nicotine)
citric acid (10%)
coffee
cooking oils
cotton seed oil
dimethyl formamide
dishwashing liquids/powders
ethyl acetate (in acetone-free nail polish remover)
ethanol (ethyl alcohol)t

ethyl ether† eucalyptol ferric chloride food colouring formalin (10% neutral buffered formaldehyde) gasoline gentian violet (crystal violet) hair dyes hemastoxlin stain household soaps hydrochloric acid (<30%) hydrogen peroxide iodine (1% in alcohol)‡ iodine, tincture of isopropanol (isopropyl alcohol)† kerosene ketchup lemon juice lipstick liquid shoe polish lye (1%) methanol[†] methyl ethyl ketone (MEK) methyl orange (1%)

as <X% were tested at multiple concentrations, with the result indicated up to the listed concentration.

A. CLASS I REAGENTS

The following reagents generally show no permanent effect on Corian[®] Solid Surface sheet when left in contact for periods of 16 hours. Wipe the surface clean using adequate personal protection for the chemical such as gloves and eye protection. Any chemical residues may be removed with a wet Scotch-Brite^{ns} pad and bleaching cleanser. Sometimes, minimal effects have been observed, particularly those indicated by footnotes (* † ‡).

methyl red (1%)	sodium sulfate
mineral oil	soy sauce
mustard	sugar (sucrose)
nail polish	sulfuric acid (<60%)
nail polish remover (acetone)	tannic acid
naphthalene (naphtha)	tea
n-Hexane	tetrahydrofuran (THF)
nitric acid (<6%)	tetramethylrhodamine
olive oil	thymol (alcohol solution)
pencil lead	toluene
perchloric acid	tomato sauce
permanent marker ink	trisodium phosphate (30%)
phenolphthalein (1%)	trypan blue
phosphorus pentoxide	urea (6%)
potassium permanganate (2%)	uric acid
povidone-iodine (PVP-I), "Betadine" Solution	urine
	vinegar
saffron	washable inks
salt (sodium chloride)	wine (all varieties)
shoe polish	Wright's stain
silver nitrate (10%)	xylenes
sodium bisulfate	zinc chloride
sodium hydroxide flaket	zinc oxide (paste, ointment)
sodium hydroxide solution (<40%)†	
sodium hypochlorite (<15%)	

* May cause surface etching or deglossing after 16 hours exposure

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May cause slight lightening after 16 hours exposure

‡ May cause slight darkening after 16 hours exposure.



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B. CLASS II REAGENTS

Corian[®] Solid Surface is not recommended for working areas where it likely to come in contact with CLASS II reagents. Concentrations reported as >X% were tested at multiple concentrations, with the result indicated above the listed concentration. The occasional stain that might result from inadvertent exposure to Class II reagents can often be removed. Scrubbing with household cleanser will remove light stains. More stubborn surface stains will require sanding with fine to coarse sandpaper, followed by typical fabrication finishing steps. Exposure to the following materials may cause damage that requires sanding for complete removal.

acetic acid (>90%) acid drain cleaners	methyl methacrylate
	methylene chloride
aqua regia	methylene chloride-based
chlorobenzene	products: paint removers, brush cleaners, some metal cleaners
chloroform (100%)	nitric acid (>25%)
cresol	phenol (>40%)
dioxane	phosphoric acid (>75%)
formic acid (>50%)	sodium hydroxide (>50%) sulfuric acid (>77%)
furfural	
hydrochloric acid 10M	trichloroacetic acid (>10%)
hydrofluoric acid (48%)	

C.2. DENTAL

Dental treatment materials may degloss, etch, or slightly stain Corian[®] surfaces. Affected areas may be restored by scrubbing with a wet Scotch-Brite[™] cleaning pad. Dental products are often proprietary blends of materials. The SDS may list some, but generally not all of the components. One common component is eugenol, which may affect the surface if not removed promptly.

Products that are not listed may be similar to the ones that are. Please compare the ingredients listed on their label or in their Safety Data Sheet (SDS) to the ones mentioned.

The published results are for 16 hours exposure time. In many cases actual exposure is much less as the material may be removed or evaporate. But, in some cases exposure can be much longer. A leaking hand-soap dispenser may cause a liquid puddle under for periods greater than 16 hours, days, or more. Similarly some containers have poorly designed spouts/caps from which product leaks every time they are used, so that they stand constantly the spilled material. If needed, a drip cup or a spill tray of a suitable material would address these situations.

The resistance to staining of Corian[®] Joint Adhesive is slightly less than that of Corian[®] Solid Surface sheet and shape.

C. SPECIALIZED PRODUCTS

C.1. BIOCHEMISTRY

Biochemistry staining agents will stain Corian[®] Solid Surface in most instances after a few minutes exposure. These stains can often be removed by prompt scrubbing with acetone. Residual stains may be restored by scrubbing with a Scotch-Brite[™] cleaning pad. Example stains are listed, but all staining agents should be handled with caution and promptly removed if spilled.

acridine orange

safranine (safranin)

gentian violet (crystal violet)

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